



1. **Calculate** the number of electrons formed in an electron avalanche which has traveled a distance of 1.5cm in the uniform field gap between two parallel plates provided that in air at the given field the values of the ionization and the attachment coefficient are $\alpha=7.4$, $\mu=5$, $\eta=2.4$ The electron avalanche has started by an electron flash of 100 electrons.
2. **Calculate** the value of secondary ionization coefficient that fulfills Townsend criterion of breakdown in a uniform gap of 2 cm width, which stressed by a uniform field corresponding to $\alpha=8$.
3. **State** Townsend criterion of breakdown in gases. In a certain gas at low pressure, if the first ionization coefficient (cm^{-1}) is related to E (volt/cm) by the expression $\alpha=(E/200)^{4.35} \cdot 10^{-6}$ and if the second Townsend coefficient has a value of 10^{-4} , **Calculate** the electrode spacing necessary to produce breakdown and the breakdown voltage assuming that E is constant at 8 kV/cm.
4. If the breakdown voltage of two parallel plates separated by a gap of 0.1cm is 4500V, **Calculate** the total secondary coefficient of ionization γ if the gap is air at a pressure 760 torr and temperature of 25° C. Take $A=15\text{cm}^{-1}$ and $B=365$.
5. If the voltage of two parallel plates separated by air gap of 0.002 m is 9 Kv just before the transition to self sustaining current. **Calculate** The total secondary ionization coefficient γ at NTP ($P=1 \text{ atm.} = 101.3 \text{ Kpa}$). The A and B values are $11253.7 \text{ (m.Kpa)}^{-1}$, 273840 (v/m.Kpa) respectively.
6. For a certain gas the first Townsend coefficient of ionization is given by the standard equation with $A=15 \text{ (cm)}^{-1}$ and $B=365$. If the secondary ionization coefficient is equal to 10^{-4} , **Calculate** the minimum breakdown voltage and the minimum value of the pressure distance product.
7. For a certain gas, if $A=15 \text{ (cm.torr)}^{-1}$ and $B=365 \text{ (v/cm.torr)}$. E/P is kept constant to be 350 V/cm.torr and P is kept constant to be 5 torr. **Calculate**:
 - a) The First Townsend's ionization coefficient α
 - b) The mean free path λ .
 - c) The ionization potential.
 - d) The maximum ionization efficiency.
 - e) If $\gamma=10^{-4}$ calculate the minimum breakdown voltage and the corresponding value of the pressure-distance product.



8. For the current growth equation of Townsend's Criterion for breakdown in Gases with second ionization process;
- Mention** the Townsend's Criterion for breakdown in Gases?
 - What** are the drawbacks of Townsend's Criterion for breakdown in Gases?
 - Define** Townsend's first and second ionization coefficients?
 - Mention** the condition for breakdown in a Townsend discharge?
 - Define** Paschen's law for breakdown in Gases?
 - Mention** how you account the breakdown voltage as a function in " $p \times d$ "?
 - Mention** how you account the minimum voltage for breakdown under a given " $p \times d$ " condition?
9. **Write** a short notes on each of the following:
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| a) Photo ionization Process. | b) Photoemission Process. |
| c) Electron attachment. | d) Self sustained discharge. |
| e) Electronegative gases. | f) Non self sustained discharge. |
| g) The various factors which affect breakdown of gases. | |
10. **Why** is the breakdown strength higher in electronegative gases compared to that in other gases?
11. **Mention** the Townsend's criterion for breakdown in electronegative gases?
12. **Explain** with drawing the streamer theory of breakdown in gases?